

This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1916 Excerpt: ...liters of a gas at 0 and 760 mm. weigh 9.9 g.; what is the molecular weight of the gas? 587. Five liters of CO₂ will have what weight when measured at 170 and 745 mm.? What will five liters weigh when measured at standard? 588. How many liters of hydrogen are in 3.30 g.? 589. Twenty-five liters of C₂H₄ at 250 and 765 mm. will have what weight? 590. Five grams of phosphorus vapor will occupy what space when measured at 5180 and 783 mm.? 591. What are the dimensions of a cube of zinc, sp. gr. 7.15, which will liberate 200 c.c. of hydrogen measured at 250 and 732 mm.? 592. Sodium phosphate, Na₂HPO₄, may be made by treating bone ash with H₂SO₄, and the resulting product, with Na₂CO₃, as follows: Ca₃(PO₄)_a + xH₂O + 2H₂SO₄ = aCaH₂(PO₄)_s + 2CaSO₄ + xH₂O, CaH₂(PO₄)_s + 2Na₂CO₃ = 2Na₂HPO₄ + aCaSO₄ + H₂O + CO₂. What weights of bone ash, sulphuric acid, and sodium carbonate are needed in the preparation of 250 g. of Na₂HPO₄? 593. If 760 g. of SiO₂ are treated with an excess of H₂F₂, what volume of SiF₄ at 180 F. and 863 mm. will result? 594. If 20 grams of NH₃ are decomposed, what volumes of nitrogen and hydrogen result at 380 F. and 700 mm.? 595. The hydrogen secured from 2 lbs. of H₂F₂ is burned; what weight of zinc is needed to liberate it, and what weight of water results? 596. Twenty cubic centimeters of water will make how many liters of steam measured at 1500 and 1300 mm.? 597. Two hundred liters of steam at 1000 and 759 mm. will make how many cubic centimeters of water when condensed? 598. Twenty-five grams of bleaching powder will yield what volume of chlorine, at 1000 F. and 743 mm., when treated with HCl? 599. To make 5 kg. of ZnCl₂, what weights of zinc and hydrochloric acid will be needed? 600. Sulphuric acid with a specific gravity of 1.674 has 75 per cent, of...

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